

# MODULE HAND BOOK

### MECHANICAL ENGINEERING VOCATIONAL EDUCATION STUDY PROGRAM

## FACULTY OF ENGINEERING – UNIVERSITAS NEGERI PADANG

COURSE NAME		CODE	Co	ourse classification	CL	J	Sem	Version		
					Theory	Pract				
Thermodynamics		MES1.61.3102	Study program o MEVE core cour	compulsory course/ se	2	0	3	1		
Responsible		Dr. Remon Lapisa, N	и.Т., M.Sc dan An	Signature						
INFORMATION		Dea	n	Head of Department	Coord	inator o	f study p	rogram		
Program Learning	Program learning outcome	Dr. Fahmi Rizal, NIP. 195912042 of Mechanical engine	1985031004	Drs. Purwantono, M.Pd NIP. 196308041986031002 education:	Drs. Purwantono, M.Pd NIP. 196308041986031002					
Outcome	<ol> <li>Possess a good abi profesional jobs / pro 1.1. possess a good problems</li> <li>possess a good u</li> <li>possess a good u</li> <li>Possess a critical and in mechanical engir <i>investigations and a</i></li> <li>problem identified</li> </ol>	ojects (Knowledge- understanding and understanding and o nderstanding and o creative thingking neering using the ssessment):	olve va chnical s techn iting va	rious teo problem ical prob	chnical 1s Ilems oblems					

	<ul> <li>2.2. problem analysis skills</li> <li>2.3. problem evaluation skills</li> <li>3. Possess a good ability in designing, manufacturing and operating machines (Engineerin 3.1. able to formulate ideas/concepts into a technical drawing, design and budget plans 3.2. able to operate various machines and other engineering equipment with the procedure</li> <li>3.3. able to design a machine or machinery system based on a valid scientific theory 3.4. able to realize a concept/design into a prototype, manufacturing process and engin</li> <li>4. Possess a good ability to design, organize and evaluate the education and learning proc vocational education. (Education design)</li> <li>4.1. able to design curriculum and learning process by considering various aspects</li> <li>4.2. able to organize, control, evaluate and improve the quality of the learning process</li> <li>4.3. able to develop an interesting, effective and efficient learning medias</li> <li>5. Possess a good ability to adapt to development in science and technology and app considering any non-technical aspects. (Engineering practice)</li> <li>5.1. able to innovate and develop technology in the field of mechanical engineering by and environmental aspects</li> <li>5.2. able to carry out the optimization process and increase the efficiency of machines of and environmental aspects</li> <li>5.2. able to carry out the optimization process and increase the efficiency of machines of and environmental aspects</li> <li>6. Possess a good oftskil and spirit of lifelong learning (Transferable skill / softskill)</li> <li>6.1. possess a religious character</li> <li>6.2. possess the ability to communicate effectively and work together in teamwork</li> <li>6.4. possess the ability to transfer science and technology to society to improve the quadi 6.5. possess a good characters of entrepreneur</li> </ul>	e correct standard operating neering system cess in <i>mechanical engineering</i> ly it into professional jobs by y considering social, economic or machining system. formation technology
Course learning	Course learning outcomes	
outcomes	CLO	PLO
	1. Understand the basic concepts of thermodynamics such as: systems and system boundaries,	1.1, 1.2, 1.3, 2.1, 2.2, 2.3
	pure substances and their characteristics, states, equilibrium, processes, cycles, the first law of thermodynamics, the second law of thermodynamics, entropy and enthalpy etc	

		d relationships between energy, heat and work and their	1.2, 1.3, 2.1, 2.2, 2.3						
	applications in mechanical engine	ering							
	3. Understand the basic concepts an	d power analysis of various power cycles (Carnot, Otto,	1.2, 1.3, 2.1, 2.2, 2.3						
	Diesel, etc.) and coolant cycles and	d their applications in mechanical engineering							
	4. Understand the basic concepts of	reactions and thermo-chemical balance in the combustion	1.2, 1.3, 2.1, 2.2, 2.3						
	process and the formation of mate	erials (fuels, gas mixtures, etc.)							
	5. able to investigate, formulate, and	alyze and apply various problems in the field of	1.2, 1.3, 2.1, 2.2, 2.3						
	thermodynamics.	modynamics.							
Course description	This course provides basic knowledge of thermodynamics which is the basis for the field of mechanical engineering and machine								
	systems, including: system concepts	and system boundaries, pure substances and their chara	cteristics, conditions, equilibrium,						
	processes, thermo-chemical reactions	, various types of cycles, the first law of thermodynamics. , t	he second law of thermodynamics,						
	entropy, enthalpy, the relationship bet	tween heat-energy and work, and their applications in mecha	nical engineering.						
References	Main Reference (RU) :								
	1. Yunus A Cengel, "Thermodynamics	s: an engineering approach", Edisi ke-2, Mc.Graw-Hills, 2014							
	2. J.P Holmann, "Thermodynamics',	McGraww-Hill International Book							
	Additional References (RP)								
	1. Bernard Challen et al, , "Diesel Engi	ne", Edisi 2 Butterworth Heinemann, 1999							
	2. Sungkono. D, "Motor Bakar Torak",	, ITS Press, 2011							
Learning media	Software:	Hardware:							
		Computer, LCD Projector and whiteboards and other devices							
Teaching team									
Assessment	Mid Test, Final Test, Individual task & t	team, group discussion and presentation							
Prerequisite courses	No								

## Course subjects

Week	Expected competencies	Topics	Method and strategy for	Assignment	Criterion / Assessment	References
	Expected competencies		leraning		indicattor	
(1)	CLO-1:	Introduction of	Subject explanation [1x80']	Students make a	Be able to explain the	RU-1
	[PLO: 1.1-2,2.1-3]	thermodynamics:	Question-answer [1x10']	summary and	concepts of	
	Students are able to	The concept of	Discussion [1x10']	description of the	thermodynamics and	
	explain the basic	thermodynamics and		presented topics in	energy, units in	
	concepts of	energy, units in		the resume book	thermodynamics,	

Week	Expected competencies	Topics	Method and strategy for leraning	Assignment	Criterion / Assessment indicattor	References
	thermodynamics.	thermodynamics, systems and conditions, temperature, pressure			systems and conditions, temperature, pressure. Score: 7%	
(2)	CLO-2: [PLO: 1.1-2,2.1-3] Students are able to explain the concept of energy, energy transfer and energy analysis (energy transfer and energy analysis)	Classification of energy (mechanical, heat, work), the first law of thermodynamics, energy conversion efficiency, energy and the environment	Review previous subject [5'] Subject explanation [1x75'] Question-answer [1x10'] Group discussion on the application of energy in a machine [1x10']	<ul> <li>students make a summary and description of the material presented in the resume book</li> <li>students discuss in groups about the application of energy in a machine</li> </ul>	Able to explain forms of energy (mechanical, heat, work), the first law of thermodynamics, energy conversion efficiency, energy and the environment. <i>Score: 7%</i>	RU-1 and RU-2
(3)	CLO-1: [PLO: 1.1-2,2.1-3] Students are able to explain the properties of pure substances	Definition and characteristics of pure substances, phases, property diagrams and phase changes, property tables, ideal gases	Review previous subject [15'] Subject explanation [1x90'] Question-answer [1x15'] Assignment [1x30']	<ul> <li>students make a summary and description of the material presented in the resume book</li> <li>students do the individual task</li> </ul>	Able to explain definitions and characteristics of pure substances, phases, property diagrams and phase changes, property tables, ideal gases <i>Score : 7%</i>	RU-1 and RU-2
(4)	CLO-2: [PLO: 1.1-2,2.1-3] Students are able to explain energy analysis in closed systems (energy analysis of closed systems)	Thermodynamic cycle in closed system energy balance, specific heat, internal energy, enthalpy for gases, fluids and solids	Review previous subject [15'] Subject explanation [1x90'] Question-answer [1x15'] Assignment [1x30']	<ul> <li>students make a summary and description of the material presented in the resume book</li> <li>students do the individual task</li> </ul>	Able to explain the energy balance in a closed system, specific heat, internal energy, enthalpy for gases, fluids and solids <i>Score : 8%</i>	RU-1 and RU-2
(5)	CLO-2: [PLO: 1.1-2,2.1-3]	Conservation of mass laws, mass balance in steady	Review previous subject [15']	<ul> <li>students make a summary and</li> </ul>	Able to explain the law of conservation of	RU-1 and RU-2

Week	Expected competencies	Topics	Method and strategy for leraning	Assignment	Criterion / Assessment indicattor	References
	Students are able to explain energy and mass analysis in open systems (energy analysis of control volume)	flow, work and energy in moving fluids, energy analysis in steady flow, turbine-compressors, heat exchangers	Subject explanation [1x90'] Question-answer [1x15'] Assignment [1x30']	description of the material presented in the resume book • students do the individual task	mass, mass balance in steady flow, work and energy in moving fluids, energy analysis in steady flow, turbine- compressors, heat exchangers. Score : 8 %	
(6)	CLO-1: [PLO: 1.1-2,2.1-3] Students are able to explain the basic concepts of the second law of thermodynamics (the second law of thermodynamics)	The basic concept of the second law of thermodynamics, thermal reservoirs, heat engines, refrigerator-heat pumps, reversible-irreversible processes, cycles and carnot engines	Review previous subject [15'] Subject explanation [1x90'] Question-answer [1x15'] Assignment [1x30']	<ul> <li>students make a summary and description of the material presented in the resume book</li> </ul>	Able to explain the basic concepts of the second law of thermodynamics, thermal reservoirs, heat engines, refrigerator- heat pumps, reversible- irreversible processes, cycle and carnot engines. Score: 7%	RU-1 and RU-2
(7)	CLO-4: [PLO: 1.1-2,2.1-3] Students are able to explain the basic concept of entropy (concept of entropy)	The concept of entropy, isentropic processes, diagrams of properties involving entropy, entropy balances	Subject explanation [1x90'] Question-answer [1x20'] Eesume scientific articles [1x40']	<ul> <li>students make a summary and description of the material presented in the resume book</li> <li>Resume scientific articles</li> </ul>	Able to explain the concept of entropy, isentropic processes, nature diagrams involving entropy, entropy balance <i>Score : 7%</i>	RU-1 and RU-2
(8)	Mid Test (UTS)					
(9)	<b>CLO-3:</b> [ <b>PLO: 1.1-2,2.1-3</b> ] Students are able to	Basics of power cycle analysis, carnot cycle, Otto cycle, Diesel Cycle	Subject explanation [1x60'] Discussion and Group discussion [1x60']	<ul> <li>students make a summary and description of the</li> </ul>	able to explain the basics of power cycle analysis, carnot cycle,	RU-1 and RU-2 RP-1 and

Week	Expected competencies	Topics	Method and strategy for leraning	Assignment	Criterion / Assessment indicattor	References
	explain the basic concepts of steam power cycle-I (gas power cycle-I)		Question-answer [1x30']	<ul> <li>material presented</li> <li>in the resume book</li> <li>Group discussions</li> <li>and presentations:</li> <li>types and</li> <li>characteristics of</li> <li>various cycles with</li> <li>reference to various</li> <li>scientific articles</li> </ul>	Otto cycle, Diesel cycle <i>Score : 7%</i>	RP-2
(10)	CLO-3: [PLO: 1.1-2,2.1-3] Students are able to explain the basic concepts of steam power cycle-II (gas power cycle- II)	Striling Cycle, Brayton Cycle, Brayton Cycle with regeneration, intercooling, jet propulsion cycle, Turbojet engine	Subject explanation [1x60'] Discussion and Group discussion [1x60'] Question-answer [1x30']	<ul> <li>tudents make a summary and description of the material presented in the resume book</li> <li>Group discussions and presentations: types and characteristics of various cycles with reference to various scientific articles</li> </ul>	Able to explain Stirling Cycle, Brayton Cycle, Brayton Cycle with regenerator, intercooling, jet propulsion cycle, turbojet engine <i>Score: 7%</i>	RU-1 and RU-2 RP-1 and RP-2
(11)	CLO-3: [PLO: 1.1-2,2.1-3] Students are able to understand the basic concepts of refrigeration cycles.	Cooling cycle and heat pump, reverse carnot cycle, analysis of the second law of the cooling cycle, selection of refrigerants, heat pumps syste	Review previous subject [3x5' = 15'] Subject explanation [1x90'] Question-answer [1x15'] Groupmassignment to explain the concept of cooling cycle [1x30']	<ul> <li>students make a summary and description of the material presented in the resume book</li> <li>Assignment: identify pump utilization in the surrounding environmentMembu at ringkasan dan diskripsi tentang</li> </ul>	Understand the types of pumps, their characteristics and utilization <i>Score: 7%</i>	RU-1 and RU-2 RP-1

Week	Expected competencies	Topics	Method and strategy for leraning	Assignment	Criterion / Assessment indicattor	References
				materi yang disampaikan dalam buku resume		
(12)	CLO-4: [PLO: 1.1-2,2.1-3] Students are able to understand the basic concepts and analysis of gas mixtures	Gas mixture composition, P-V-T chart for gas mistures, properties of gas mistures	Review previous subject [3x5' = 15'] Subject explanation [1x105'] Question answer [1x30']	<ul> <li>students make a summary and description of the material presented in the resume book</li> </ul>	understand gas mixture composition, P-V-T chart for gas mistures, properties of gas mixtures <i>Score: 7%</i>	RU-1 and RU-2
(13)	CLO-4: [PLO: 1.1-2,2.1-3] Mahasiswa mampu memahami konsep dasar reaksi kimia termodinamika-1 (chemical reaction-1)	Fuel and combustion, theoretical and actual combustion processes, enthalpy formation and combustion	Review materi oleh mahasiswa [3x5' = 15'] Penjelasan materi [1x105'] Diskusi dan tanya jawab [1x30']	<ul> <li>students make a summary and description of the material presented in the resume book</li> </ul>	Fuel and combustion, theoretical and actual combustion processes, enthalpy formation and combustion <i>Scoret: 7%</i>	RU-1 dan RU-2 RP-1
(14)	CLO-4, 5: [PLO: 1.1-2,2.1-3, 6.3] Students are able to understand the basic concepts of thermodynamic chemical reactions-2 (chemical reaction-2)	Analysis of law I of thermodynamic reaction systems, adiabatic flame temperature, changes in entropy of change systems, law 2 of reaction system analysis, criteria for chemical equation reactions, phase equilibrium,	Group presentation [1x120'] Question answer [1x20'] Conclusion [1x10']	<ul> <li>Preparing group presentation material on non- conventional energy (4 groups)</li> <li>Group discussion on renewable energy technologies</li> </ul>	Able to analyze first law of thermodynamic reaction systems, adiabatic flame temperature, changes in entropy of change systems, law 2 of reaction system analysis, chemical equation reaction criteria, equilibrium phase <i>Scoe: 7%</i>	RU-1 and RU-2
(15)	CLO-2, 4, 5: [PLO: 1.1-2,2.1-3, 6.3]	Stagnation properties, sound velocity and number	Group presentation [1x120']	<ul> <li>Preparing group presentation</li> </ul>	Able to explain the nature of stagnation,	RU-1 and RU-2

Week	Expected competencies	Topics	Method and strategy for leraning	Assignment	Criterion / Assessment indicattor	References
	Compressible flow	March, 1D isentropic flow, isentropic process flow in Nozzles, shock and expansion waves, Steam Nozzles	Question answer [1x20'] Conclusion [1x10']	<ul><li>material on</li><li>Compressible flow</li><li>(4 groups)</li><li>Group discussion on compressible flow</li></ul>	sound velocity and March number, 1D isentropic flow, isentropic flow process in Nozzles, shock and expansion waves, Steam Nozzles Score : 7%	
(16)	Final Test (UAS)					

<u>**Remark**</u>: 1 CU = (50' TM + 60' BT + 60' BM)/week TM = Classical session BT = Assignment.

BM = Individual session

PS = Simulation practice (160 menit/week)

T = Theory (cognitif aspects)

P = Practicum (psychomotoric aspects)

PL = Laboratory activites (160 menit/week)

#### Relationship between CLO and PLO and assessment methods

Wee k	MES1.61.3102	Assessment	Quality (%)	PLO-1			PLO-2		PLO-3			PLO-4			PLO-5			PLO-6						
				1	2	3	1	2	3	1	2	3	4	1	2	3	1	2	3	1	2	3	4	5
1	CLO-1	UTS.1	5	V	V	V	V	V	V															
2	CLO-2	UTS.2	5	V	V	V	V	V	V															
3	CLO-1	UTS.3	5	V	V	V	V	V	V															
4	CLO-2	UTS.4	5	V	V	V	V	V	V															
5	CLO-2	UTS 5	5	V	V	V	V	V	V															
6	CLO-1	UTS 6	5	V	V	V	V	V	V															
7	CLO-4	UTS 7	5	V	V	V	V	V	V															
9	CLO-3	UAS 1	7	V	V	V	V	V	V															
10	CLO-3	UAS 2	7	V	V	V	V	V	V															
11	CLO-3	UAS 3	7	V	V	V	V	V	V															

12	CLO-4	UAS 4	7	V	V	V	V	V	V										
13	CLO-4	UAS 5	7	V	V	V	V	V	V										
14	CLO-4, 5	Presentation	10	V	V	V	V	V	V									V	
15	CLO-4, 5	Presentation	10	V	V	V	V	V	V									V	
16	CLO-1-5	Assignment	15	V	V	V	V	V	V	V	V	V	V						
	TOTAL		100																

#### Assessment components

Mid Test (UTS)	: 30 %
Final Test (UAS)	: 30 %
Student assignment	: 30 %
Presence	: 10 %
Total	: 100 %

## Scoring/Grading level description

	Excellent	Good	Satisfy	Fail
ability to describe	Able to describe correctly	Able to describe correctly	Able to describe but less	Unable to describe
	and completely	but not complete	clear and incomplete	
ability to formulate	Able to formulate correctly	Able to formulate correctly	Able to formulate but less	Unable to formulate
	and completely	but not complete	clear and incomplete	
ability to calculate	Able to calculate correctly	Able to calculate correctly	Able to calculate but less	Unable to calculate
	and completely	but not complete	clear and incomplete	
ability to analyze	Able to analysize correctly	Able to analyze correctly but	Able to analyze but less clear	Unable to analyze
	and completely	not complete	and incomplete	

## Scoring and grading system

Score	Quality	Quality score	Designation	Score	Quality	Quality score	Designation
85 – 100	А	4.0	Outstanding	55 – 59	С	2.0	Acceptable
80 - 84	A-	3.6	Excellent	50 - 54	C-	1.6	Poor
75 – 79	B+	3.3	Very good	40 - 49	D	1.0	Poor
70 – 74	В	3.0	Good	≤ 39	E	0.0	Fail
65 - 69	B-	2.6	Good	-	Т	-	Tertunda
60 - 64	C+	2.3	Acceptable				